# TRILOGY

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### URIC ACID

## Quantitative Determination of Uric Acid Enzymatic Colorimetric Uricase POD Monoreagent

For professional in vitro diagnostic use only.

Stability in urine: 4 days at 15° - 25°C, Discard contaminated specimens

#### TEST PROCEDURE

Wavelength : 505 nm Light path : 1 cm Temperature : 37 °C

Measurement: : Against Reagent Blank

Assay type : ENDPOINT

#### Assay:

	Blank	Assay
Reagent	1000 μL	1000 μL
Sample/ Standard/ Cal	1	20 μL

- -Mix and incubate for 5 min at 37 °C or for 10 min at room temperature (16-25 °C).
- Read the absorbance (A) of the samples and the standard against the reagent blank.

#### CALCULATION

#### Serum/ Plasma:

Uric Acid Concentration= Abs Sample - Abs blank

Abs Std/Cal - Abs blank x Std/ Cal.

#### Urine

Calculate as for the serum and multiply the result by 10 (initial sample dilution).

Uric Acid (mg/24 h) = Uric Acid (mg/dL) x Urine Vol. 24 h (dL). Uric Acid ( $\mu$ mol /24 h) = Uric Acid ( $\mu$ mol /L) x Urine Vol. 24 h (L). Conversion Factor: [mg/dL] x 59.48 = [ $\mu$ mol/L]

#### **QUALITY CONTROL**

Normal and abnormal control sera of known concentration should be analyzed routinely with each group of unknown samples.

#### **EXPECTED VALUES**

Serum/plasma:

Male:

• 3.5 -7.2 mg/dL.

Female:

• 2.6 - 6.0 mg/dL.

Urine(24h):

- $\leq$  800 mg/24h (4.76 mmol/24h) assuming normal diet
- $\leq$  600 mg/24h (3.57 mmol/24h) assuming low purine diet.

Each laboratory should establish a range of expected values based on its patient population and, if necessary, determine its own reference interval. For diagnostic purposes, results should always be assessed together with the patient's medical history, clinical examination and other results.

#### INTENDED USE

Quantitative determination of Uric Acid in Serum, Plasma or Urine. Enzymatic Colorimetric method (Uricase-POD)

#### **GENERALITIES**

Uric acid is the major product of the catabolism of purine nucleosides (adenosine and guanosine) from the purine metabolism pathway. Purines may be synthesized endogenously from the breakdown of nucleic acids or may be obtained from sources as diets in which nucleic acids are present. An abnormal increase in the level of uric acid in circulation above 7.0 mg/dL (0.42 mmol/L) is referred to as hyperuricemia, being the gout the major form of the ailment resulting in the deposition of urates in the soft tissues, especially in the joint areas. Increased levels may be also found associated with leukemia, toxemia of pregnancy and severe renal impairment. Less common are the cases of hypouricemia where the concentration of uric acid is below 2.0 mg/dL (0.12 mmol/L). These cases are usually secondary to cases of hepatocellular disease, renal reabsorption defect, or overtreatment with uricosuric drugs used in the treatment of hyperuricemia.

#### **TEST PRINCIPLE**

Uricase oxidizes Uric Acid to Allantoin with the formation of Hydrogen Peroxide which, in the presence of Peroxidase (POD), reacts with a chromogen (DHBS) and 4-Amino-Antipyrin, forming a red colored compound whose intensity, measured photometrically at 505 nm, is proportional to the concentration of Uric Acid in the sample.

#### REAGENT COMPOSITION

#### Reagent

 Good's buffer, pH 7.8
 100 mmol/L

 Uricase
 > 0.5 KU/L

 Peroxidase (POD)
 > 0.5 KU/L

 4-aminoantipyrine
 0.5 mmol/L

 DCPS
 2 mmol/L

Standard

Uric Acid value on label

#### STORAGE, PREPARATION AND SHELFLIFE

Liquid and ready to use reagents, stable until the expiry date shown, if stored as indicated on the label and avoid contamination, evaporation and prolonged exposure to direct light. Do not freeze the reagents. Discard the reagent if signs of deterioration appear, such as the presence of particles and turbidity or failure to recover the values of certified control sera. After opening the bottles, it is advisable to withdraw the necessary volume, to immediately close the bottles and store them in the fridge to avoid contamination, degradation from direct light and evaporation. The measurement is not influenced by occasionally occurring color changes, as long as the absorbance of the reagent is < 0.1 at 505 nm.

#### **SAMPLE COLLECTION AND PREPARATION**

Serum, Heparin or EDTA plasma, Urine 24h diluted 1:10 with H2O (1+9).

Stability in serum/plasma: 7 days at 4°-8° C. 3 months at - 20°C Only freeze once. Discard contaminated specimens.

#### **PERFORMANCE**

#### **PRECISION:**

Low Level: Samples= 20; Average = 5.6; S.D. = 0.04; CV = 0.70%. High Level: Samples = 20; Average = 9.14; S.D. = 0.04; CV = 0.41%.

#### **ACCURACY:**

A comparison between this method (x) and a certified method of trade (y) gave the following correlation:

y=1.03x r=0.978

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**SENSITIVITY:** 0.03 mg/dL. **LINEARITY:** 30 mg/dL.

#### **PRECAUTIONS**

R contains PHENOL, 4-AMINOANTIPYRINE.

in case of contact of reagents with the operator, you must apply the following first aid:

In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. After contact with skin, wash immediately with plenty of water. Wear suitable protective clothing, gloves and eye/face protection.

#### **INTERFERENCES**

No interference was observed by Ascorbic Acid up to 20 mg/dL, Hemoglobin up to 200 mg/dL, Bilirubin up to 10 mg/dL and lipemia up to 2000 mg/dL Triglycerides.

#### **LITERATURE**

- Barham, D. and Trinder, P. Analyst. 97: 142 (1972)
- Tietz. N.W. Clinical Guide to Laboratory Tests, 3rd Edition. W.B. Saunders Co. Philadelphia, PA. (1995).
- Young DS. Effects of drugs on clinical laboratory tests, 5th ed. AACC Press, 2000.
- Fossati, P., Prencipe, L. and Berti, Q. Clin. Chem. 26: 227 (1980).

#### **USED SYMBOLS**

IVD	In Vitro Diagnostic Medical Device
••••	Manufacturer
~··	Date of Manufacture
REF	Catalogue Number
LOT	Batch Code
$\subseteq$	Use by YYYY-MM (MM = end of month)
[]i	Operator's Manual; Operating Instructions
*	Keep away from Sunlight
<del>*</del>	Keep away from Rain
X	Temperature Limit
$\triangle$	Caution
<b>(See )</b>	Do not use if Package is Damaged
2	Do Not Re-Use
$\overline{\Sigma}$	Contains Sufficient for <n> Tests</n>

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